Man Made Substances

- **1. Fertilizers:** The substances added to the soil to make up the deficiency of essential elements are known as fertilizers, these are either natural or synthetic (chemical). For a chemical fertilizer, the following requirements should be met:
- (i) It must be sufficiently soluble in water
- (ii) It should be stable so that the element in it may be a ailab e for a longer time.
- (iii) It should contain nothing injurious to plants.

Among the chemical fertilizers the two import nt categories are :

Phosphatic Fertilizers: All naturally occu ing phosphates are orthophosphates, the most abundant of these being rock phosphate [Ca3(PO4)2], which is mostly con um d by the fertilizer industry in the manufacture of 'superphosphate of ime' triple superphosphate' and 'nitrophos'— a combined pho phatic and nitrogenous fertilizer. Other phosphatic fertilizers e ammoninum dihydrogen orthophosphate and d ammonium hydrogen orthophosphate, which also conteract nitrogen ficiency.

Nitrogenous Fe tilizers: P ants need nitrogen for rapid growth and increase in their rotein c ntent. For this reason, nitrogenous fertilizers be ome more important. The chief nitrogeneous fertilizers are am onium sulpha e, calcium cyanamide, sodium nitrate, ammoniu n te urea, diammo-nium phosphate and ammonium phosphate.

2. Dyes : Coloured substances used for colouring textiles, foodstuffs, silk, wool, e are called dyes.

Different classes of dyes are given below.

(i) **Nitro dyes:** These are polynitro derivatives of phenol where nitro group acts as a chromophore and hydroxyl group as auxochrome. These are less important industrially because the colours are not fast.

- (ii) Azo dyes: These are an important class of dyes and are characterised by the presence of azo group (-N=N-) as the chromophore. The groups like NH2, NR2 or -OH, etc., present in the molecule containing one or more azo gruops act as the auxochromes.
- (iii) Tripheny Imethane dyes: These dyes contain the par aquinoid moiety as a chromophore and —OH, —NH2 or —NR2 as auxochrome. These dyes are not fast to light and washing and hence are mainly used for colouring paper or typewriter ribbons, e.g. malachit green which is used for dyeing wool and silk directly and cotton after mordanting with tannin.
- (iv) Mordant dyes: Those dyes which are fixed on the fibre that the help of a mordant are known as mordant dyes. Fo acidic dyes, basic mordants (such as hydroxides of iron, aluminium a dichromium) are used, while for basic dyes, acidic mordants (i e tanneacid) are used. Here the fabric is first dipped into a solution of mordant and then into the dye solution. The colour produced depends on the nature of the mordant used.
- (v) Vat dyes: These are water insoluble dy s and are introduced into the fibre in its (soluble) reduced orm, also known as leuco form (colourless). These are call divated es because reducing operation (using sodium hydros liphite) with sformerly carried out in wooden vats. Indigo is a vatidye and lused for dyeing cotton.

Cement: It is a omplex m erial containing the silicates of calcium and aluminium. A paste o it in water sets into a hard rocky mass-called the setting f cement. A paste of sand, cement and water called mortar is ve y conve ently used for joining bricks and plastering wa s.

A mixtu e of s one chips (gravel) sand cement and water, known as concrete. ets harder than ordinary mortar. It is used for flooring and making road Concrete with steel bars and wires called reinforced concrete (RC) forms a very strong material. It is used for constructing roofs, bridges and pillars.

Glass : Supercooled liquid is called glass. SiO2 is it's common constituent.

- (a) Soda glass or soda lime glass: It is Sodium calcium silicate (Na2O CaO 5 SiO2). It is the cheapest of all glasses and used for making window panes and bottles and easily attacked by chemicals.
- **(b) Potash glass:** It contains potassium in place of sodium, it has higher softening temperature as also a greater resistance to chemicals. So used for chemical apparatus; beakers, flasks, funnels etc.
- (c) Optical glass: It is used for making lenses, prisms and optical instruments like telescopes and microscopes. It contains boric o ide (B2O3) and silica (SiO2)

Types: (i) Crown glass: Contains K2O & BaO a the basi ide

- (ii) Flint glass: Contains PbO as the basic xid
- (d) Crooks glass: for spectacles: Absorbs ultraviol t rays which are harmful for the eyes.
- **(e) Lead crystal and crystal glass** Lead g ss sparkles used for making decorative items. It cont ins 2 % or more of PbO called lead crystal. If it contains term than 4% lead oxide called crystal glass.
- **(f) Borosilicate gla s:** t co tains ess alkali (K2O or CaO3) and more SiO2 than potash g ass and some B2O
- **g) Coloured glas**: It is u ed for making artificial jewellery, crockery and stained glass window .
- **(h) Mil y g ass :** Milky glass is prepared by adding tin oxide (SnO2). Calcium pho pha (Ca3(PO4)2) or cryolite (AH33NaF) to the melt glass. If the e substances are white so look milky.
- (i) Glass I minates: It is made by fixing polymer sheets between layers of glas It is used to make windows & Screens of cars, trains and aircraft specially manufactured glass laminates are used bulletproof material.

Some common man-made polymers and their uses.

Paints: Chemical, contains a pigment as a vehicle and a thinner.

White pigment: Zinc oxide, white lead and titanium dioxide. The pigment is mixed with a vehicle, which is an oil like linseed or soyabean oil or a polymer. A thinner is a solvent such as turpentine oil or kerosene.

Luminous paints : Glow when exposed to light. Paints are applied on a surface to protect it from corrosion and weathering or to give it an attractive look.

Soaps and Detergents : Soaps are the sodium or Potass um s ts of fatty acids. They are made by the saponification of fats. Detergent are made from some petroleum products.

Antibiotic : Medicinal compounds produced by moulds a d bacteria, capable of destroying or preventing the growth of acteria in animal systems.

Antibody: Kinds of substances formed in he bl od, tending to inhibit or destroy harmful bacteria, etc.

Antidote: Medicine used agains a po on, o to prevent a disease from having effect.

Antigen: Substance capable f stim lating formation of antibodies.

Antimony: A brittle cry talline, silvery white metal.

Antipyretie: A ubstanc used to lower body temperature.

Pesticides Many iving organism destroy crops or eat away grains. They ar col ectively known as pests. To kill chemical used called pesticides

Insectic des : D.D.T. aluminium phosphate gammexine.

Fungicide: hiram, Bordeanx mixture CaCaSO45H2O + (OH)2

Rodenticides: Aluminium phosphide.

Herbicides : Benzipram, benzadox.

Medicines: To cure diseases by biological changes in the body.

Analgesics : Painkillers are called analgesics eg, Aspirin, Paracetamol and morphine.

Antimalarial drugs: Used to treat malaria quinine derivatives eg, chlovoquine.

Destroy microorganism : Penicillin, Aminogly considers, oftoxaim, Homophonic.

Sulphadrugs: Alternatives of antibiotics, sulphanilamide sulphadiazine, Sulpha gunamidine.

Antaoxide: Substances which remove the excess aci and is the pH to appropriate level in scotch are called antacids. It is caused by excess of HCl in the gastric juice magnesium hydre, mag zines carbonate, magnesium truistical, aluminium phosphe e are common antacids.

Epsom salt : Hydrated magnesium sulphate MgSO4 · 7H2O), used in medicines to empty bowels.

Chloroform: A sweetish, colou ess liquid. It is used as a solvent and anaesthetic.

Saccharin: A white cr talline s lid which is 550 times sweeter than sugar, but does not have ny food value. It is used by diabetic patients.

DDT: Dichloro d henyl t choloro ethane, a white powder used as an insecticide.